Organic Synthesis

Intermolecular Tandem Pd-Catalyzed Cross-Coupling/[4+4] and [4+2] Cycloadditions: A One-Pot, Five-Component Assembly of Bicyclo[6.4.0]dodecanes**

Phil Ho Lee* and Kooyeon Lee

Tandem reactions have long been established as efficient methods for the rapid synthesis of complex compounds starting from simple, readily available substrates in relatively few steps.^[1-3] Although many intramolecular tandem reactions have been reported for constructing complex organic molecules, the corresponding highly efficient intermolecular reactions remain an important challenge for synthetic organic chemists. A range of strategies involving the sequential generation of radical and anionic species has been used for such intermolecular transformations. [4,5] However, relatively few transition-metal-catalyzed intermolecular tandem processes^[6] have been developed for the synthesis of complex cyclic compounds.^[7] Recently, we reported a method for the synthesis of substituted allenes, polyallenes, and unsymmetrical bis(allenes) from allenylindium reagents generated in situ.[8] With this result in hand, we envisioned that a Pdcatalyzed cascade process involving sequential cross-coupling between vinyl bromides or triflates and propargyl halides followed by [4+4] and [4+2] cycloadditions should provide structurally diverse carbocycles. Herein, we describe the development of tandem Pd-catalyzed cross-coupling/[4+4]

^[*] Prof. Dr. P. H. Lee, Dr. K. Lee Department of Chemistry, Kangwon National University Chunchon 200-701 (Republic of Korea) Fax: (+82) 33-253-7582 E-mail: phlee@kangwon.ac.kr

^[**] This work was supported by the CMDS at KAIST and grant no. R02-2003-000-10023-0 of the Basic Research Program of the KOSEF. We thank Prof. Tom Livinghouse of Montana State University for proofreading this manuscript.

Zuschriften

and [4+2] cycloadditions as an efficient approach to bicyclo-[6.4.0]dodecane derivatives (Scheme 1).^[9]

Scheme 1. EWG = electron-withdrawing group.

The tandem process was first examined with α-bromostyrene, propargyl bromide, and tetracyanoethylene (TCE). The organoindium reagent obtained from one equivalent of indium and 1.5 equivalents of propargyl bromide was added to a solution of one equivalent of α -bromostyrene in the presence of 4 mol % of [Pd(PPh₃)₄] and three equivalents of LiCl, [8,10] and the solution was stirred at 50 °C for 2 h. In the ¹H NMR spectrum of the crude product mixture, 3,4dimethylene-1,6-diphenyl-1,5-cyclooctadiene observed as the major compound, thus indicating that a sequential cross-coupling/[4+4] cycloaddition had taken place to produce 4ba in 94% yield. Because 4ba is a 1,3diene, we further attempted a [4+2] cycloaddition reaction with crude 4ba. Thus, 4ba was treated with 1.5 equivalents of TCE (70°C, benzene, 18 h) to afford 6baa in 84% yield. This transformation shows that five components can be assembled in an intermolecular cascade cross-coupling/[4+4] and [4+2] cycloaddition sequence (Scheme 2).

A Pd-catalyzed [4+4] cycloaddition reaction with a vinyl allene as the starting material was reported recently. ^[11] In this reaction, the vinyl allene was obtained from the reaction of 1-phenylvinylmagnesium bromide with propargyl bromide in the presence of a Pd⁰ catalyst. ^[12] However, preparation of the vinyl allene in situ by inversion of the charge polarization of the reaction components (α -bromostyrene and propargyl bromide) was applied in the present method.

Scheme 2.

The results of several one-pot, five-component assembly reactions are summarized in Table 1. With 5b, 5c, 5e, and 5g as dienophiles, 4ba gave bicyclo[6.4.0]dodecane derivatives in good yields (entries 2-5). Similarly, 4ba reacted with maleimide (5 f) to produce 6baf in 79% yield (entry 6). For a vast number of α-bromovinylarenes as organic electrophiles, the presence of various substituents, for example, 4-trifluoromethyl, 4-methoxy, and 2-chloro on the aromatic ring, did not affect the efficiency of the tandem reactions. The reaction also worked equally well with α -bromovinylarenes containing free hydroxy and amino groups (entries 9 and 11). Treatment of 4ga with two equivalents of naphthoquinone afforded 6gad in 64% yield (entry 7). Reaction of 4da with glyoxylic acid ethyl ester produced 6dah in 69% yield, and the enol triflate of 3bromoacetophenone underwent the cascade reaction to afford **6hae** in 66% yield (entry 12).

(α-Bromovinyl)trimethylsilane reacted with propargyl bromide to produce $\bf 4aa$ in 71% yield under the optimized conditions. Unfortunately, the desired bicyclo[6.4.0]dodecane derivative was not obtained, because of the instability of $\bf 4aa$ under the conditions of the [4+2] cycloaddition reaction. With propargyl halides as nucleophilic cross-coupling partners, the presence of a methyl substituent at the α position has little effect on either the reaction rate or the product yield. In the case of 3-chloro-3-methyl-1-butyne ($\bf 2b$), $\bf 4bb$, $\bf 4cb$, $\bf 4eb$, and $\bf 4gb$, with four methyl substituents on the exocyclic C=C bonds, were obtained in 63–91% yield. However, these compounds did not undergo the [4+2] cycloaddition because of steric hindrance.

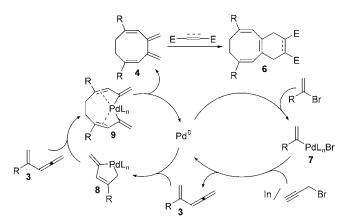
Although the mechanism of the present reaction has not been established, a possible reaction pathway is shown in

Table 1: Cascade cross-coupling/[4+4] and [4+2] cycloaddition. [a]

Entry	Reagents	Product	Yield [%] ^[b]	Entry	Reagents	Product	Yield [%] ^[b]	Entry	Reagents	Product	Yield [%] ^[b]
1	1 b 2 a 5 a	Ph CN CN CN CN Ph 6baa	84	5	1 b 2 a 5 g	CO ₂ Me CO ₂ Me	74	9	1 e 2 a 5 a	3-H ₂ N-C ₆ H ₄ CN CN CCN 3-H ₂ N-C ₆ H ₄ 6eaa	65
2	1 b 2 a 5 b	Ph CO ₂ Me CO ₂ Me	76	6	1 b 2 a 5 f	Ph O NH O O O O	79	10	1 f 2 a 5 c	2-Cl-C ₆ H ₄ O	72 ^[c]
3	1 b 2 a 5 c	Ph O O Ph 6bac	72 ^[c]	7	1 g 2 a 5 d	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	64	11	1 d 2 a 5 h	3-HO-C ₆ H ₄ CO ₂ Et	69
4	1 b 2 a 5 e	Ph CO ₂ Et	71	8	1 c 2 a 5 c	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	79 ^[c]	12	1 h 2 a 5 e	$3\text{-Br-C}_{6}\text{H}_{4} \qquad \qquad \text{CO}_{2}\text{Et}$ $3\text{-Br-C}_{6}\text{H}_{4} \qquad \textbf{6hae}$	66

[a] Cross-coupling/[4+4] cycloaddition: Compound 1 (0.5 mmol) was treated with 2 (0.75 mmol) and indium (0.5 mmol) in THF (2 mL), in the presence of 4 mol% of [Pd(PPh₃)₄] and LiCl (1.5 mmol) under N_2 at 50°C for 2 h. [4+2] Cycloaddition: Crude 4 was treated with 5 (0.38 mmol) in benzene (2 mL) at 70°C for 18 h. [b] Yield of isolated product. [c] 2.5 equiv of methyl vinyl ketone was used.

TMS R 4bb: R = Ph, 91% 4cb: R = 4-MeO-
$$C_6H_4$$
, 79% 4eb: R = 3- H_2N - C_6H_4 , 63% 4gb: R = 4- F_3C - C_6H_4 , 71%



Scheme 3.

Scheme 3. Oxidative addition of the vinyl bromide to a Pd⁰ complex and subsequent transmetalation with the organoindium reagent, followed by reductive elimination, affords the vinylallene **3**. Subsequent insertion of Pd⁰ into the vinyl allene produces a five-membered palladacycle **8**, which reacts with another molecule of the vinyl allene to generate the di(σalkenyl) palladium complex **9**. Final reductive elimination would then give the 3,4-dimethylene-1,5-cyclooctadiene **4**.^[8,11] [4+2] Cycloaddition of the 1,3-diene **4** with representative dienophiles produces the observed bicyclo[6.4.0]dodecane derivatives **6**.

In conclusion, this study has led to the development of a novel tandem Pd-catalyzed cross-coupling/[4+4] and [4+2] cycloaddition sequence that allows the rapid synthesis of bicyclo[6.4.0]dodecane derivatives starting from $\alpha\text{-bromovinylarenes}, propargyl bromides, and dienophiles in one reaction vessel. It is noteworthy that five components are assembled into one molecule in this procedure. In addition, the present process is one of the comparatively few examples in which a <math display="inline">Pd^0$ catalyst is simultaneously involved in two catalytic cycles. $^{[2,3]}$

Experimental Section

Typical experimental procedure: 6baa: α-Bromostyrene (1b, 102.0 mg, 0.5 mmol) was added at room temperature, under nitrogen, to a suspension of [Pd(PPh₃)₄] (4 mol%, 23.1 mg) and lithium chloride (63.5 mg, 1.5 mmol) in dry THF (1 mL). After 15 min the allenylindium reagent, which was generated from propargyl bromide (2a, 80 % (w/w) in toluene, 89.2 mg, 0.75 mmol) and indium (57.0 mg, 0.5 mmol) in dry THF (1 mL), was added. The solution was stirred at 50 °C under nitrogen for 2 h, and then the solvent was removed under reduced pressure. A solution of tetracyanoethylene (5a, 48 mg, 0.38 mmol) in dry benzene (2 mL) was added to the residue. After heating at 70°C under nitrogen for 18 h, the resulting solution was quenched with saturated aqueuous NaHCO3. The aqueous layer was extracted with diethyl ether (3×20 mL), and the combined organic phases were washed with water and brine, dried with MgSO₄, filtered, and concentrated under reduced pressure. The residue was purified by silica-gel column chromatography with EtOAc/hexane (1:10) as eluent to produce 6,9-diphenyl-7,8-dihydro-1H,4H-benzocyclooctene-2,2,3,3-tetracarbonitrile (**6baa**; 87.0 mg, 84%). ¹H NMR (400 MHz, CDCl₃, 25 °C, TMS): $\delta = 7.34-7.30$ (m, 10 H), 5.94 (s, 2H), 3.28 (s, 4H), 2.88 ppm (s, 4H); 13 C NMR (100 MHz, CDCl₃): $\delta =$ 147.3, 141.9, 128.6, 128.3, 126.2, 124.2, 124.0, 110.6, 37.9, 36.6,

Zuschriften

30.4 ppm; IR (film): $\tilde{\nu}$ = 3055, 1941 cm⁻¹; HRMS (EI) calcd for $C_{28}H_{20}N_4M^+$ 412.1688; found 412.1689.

Received: September 12, 2004 Revised: February 11, 2005 Published online: April 21, 2005

Keywords: cross-coupling \cdot cycloaddition \cdot indium \cdot palladium \cdot tandem reactions

- [1] N. Hall, Science 1994, 266, 32.
- [2] a) T.-L. Ho, Tandem Organic Reactions, Wiley-Interscience, New York, 1992; b) T.-L. Ho, Tactics of Organic Synthesis, Wiley-Interscience, New York, 1994, p. 79; c) L. F. Tietze, F. Haunert in Stimulating Concepts in Chemistry (Eds.: F. Vögtle, J. F. Stoddart, M. Shibasaki), Wiley-VCH, Weinheim, 2000, p. 39.
- [3] a) L. F. Tietze, U. Beifuss, Angew. Chem. 1993, 105, 137; Angew. Chem. Int. Ed. Engl. 1993, 32, 131; b) R. A. Bunce, Tetrahedron 1995, 51, 13103; c) E. Negishi, C. Coperet, S. Ma, S.-H. Liou, F. Liu, Chem. Rev. 1996, 96, 365; d) L. F. Tietze, Chem. Rev. 1996, 96, 115.
- [4] a) I. Ryu, H. Yamazaki, A. Ogawa, N. Kambe, N. Sonoda, J. Am. Chem. Soc. 1993, 115, 1187; b) K. Tsuchii, M. Doi, T. Hirao, A. Ogawa, Angew. Chem. 2003, 115, 3614; Angew. Chem. Int. Ed. 2003, 42, 3490.
- [5] a) T. Shono, I. Nishiguchi, M. Sasaki, J. Am. Chem. Soc. 1978, 100, 4314; b) K. Takai, T. Ueda, N. Ikeda, T. Moriwake, J. Org. Chem. 1996, 61, 7990; c) K. Takai, N. Matsukawa, A. Takahashi, T. Fujii, Angew. Chem. 1998, 110, 160; Angew. Chem. Int. Ed. 1998, 37, 152; d) J. Terao, K. Saito, S. Nii, N. Kambe, N. Sonoda, J. Am. Chem. Soc. 1998, 120, 11822.
- [6] a) M. Lautens, W. T. Klute, Chem. Rev. 1996, 96, 49; b) L. Yet, Chem. Rev. 2000, 100, 2963; c) S.-I. Ikeda, Acc. Chem. Res. 2000, 33, 511; d) J. Montgomery, Acc. Chem. Res. 2000, 33, 467; e) Y. M. Dong, W. C. MacMillan, J. Am. Chem. Soc. 2001, 123, 2448; f) P. A. Wender, G. G. Gamber, M. J. C. Scanio, Angew. Chem. 2001, 113, 4013; Angew. Chem. Int. Ed. 2001, 40, 3895; g) S. Ikeda, J. Synth. Org. Chem. Jpn. 2001, 59, 960; h) L. W. A. van Berkom, G. J. T. Kuster, F. Kalmoua, R. de Gelder, H. W. Scheeren, Tetrahedron Lett. 2003, 44, 5091; i) K. Inanaga, K. Takasu, M. Ihara, J. Am. Chem. Soc. 2004, 126, 1352; j) A. Ajamian, J. L. Gleason, Angew. Chem. 2004, 116, 3842; Angew. Chem. Int. Ed. 2004, 43, 3754.
- [7] a) T. K. Devon, A. I. Scott, Handbook of Naturally Occurring Compounds, Vol. II, Academic Press, New York, 1972; b) D. J. Faulkner, Nat. Prod. Rep. 1984, 1, 251, 551; D. J. Faulkner, Nat. Prod. Rep. 1986, 3, 1; D. J. Faulkner, Nat. Prod. Rep. 1987, 4, 539; D. J. Faulkner, Nat. Prod. Rep. 1988, 5, 613; c) T. Oishi, Y. Ohtsuka, Studies in Natural Products Chemistry (Ed.: Atta-ur-Rahman), Elsevier, Amsterdam, 1989, p. 73; d) C. J. Moody, Studies in Natural Products Chemistry (Ed.: Atta-ur-Rahman), Elsevier, Amsterdam, 1992, pp. 201 239.
- [8] K. Lee, D. Seomoon, P. H. Lee, Angew. Chem. 2002, 114, 4057; Angew. Chem. Int. Ed. 2002, 41, 3901.
- [9] For a review on eight-membered ring carbocycle construction, see: a) N. A. Petasis, M. A. Patane, *Tetrahedron* 1992, 48, 5757;
 b) S. M. Sieburth, N. T. Cunard, *Tetrahedron* 1996, 52, 6251;
 c) G. Mehta, V. Singh, *Chem. Rev.* 1999, 99, 881.
- [10] a) P. H. Lee, S.-Y. Sung, K. Lee, Org. Lett. 2001, 3, 3201; b) K. Lee, J. Lee, P. H. Lee, J. Org. Chem. 2002, 67, 8265; c) P. H. Lee, D. Seomoon, K. Lee, S. Kim, H. Kim, H. Kim, E. Shim, M. Lee, S. Lee, M. Kim, M. Sridhar, Adv. Synth. Catal. 2004, 346, 1641.
- [11] a) M. Murakami, K. Itami, Y. Ito, Angew. Chem. 1998, 110, 3616;
 Angew. Chem. Int. Ed. 1998, 37, 3418; b) M. Murakami, K. Itami,
 Y. Ito, Synlett 1999, 951; c) Thermal dimerization of vinylallene:

- R. Schneider, H. Siegel, H. Hopf, *Liebigs Ann. Chem.* 1981, 1812
- [12] When α-bromostyrene is treated with propargyl bromide in the presence of Mg and Pd⁰, α-bromostyrene and propargyl bromide act as synthons of a vinyl anion and allenyl cation, respectively, to produce a vinyl allene: M. Murakami, K. Thami, Y. Ito, Organometallics 1999, 18, 1326.

[13] The use of 3-bromo-1-butyne produced the corresponding [4+4] adduct in 90% yield. The product consisted of three stereo-isomers (1.3:1.0:1.3) with respect to the orientation of the two methyl groups on the exocyclic double bonds.